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## Investigation Of Machinability Characteristics Of Sae Ams4413b Aluminum Alloy Using Ethylene Glycol As A Coolant

Roopa K. Rao1\*, Sachin C. Kulkarni<sup>1</sup>, Shivakumar S<sup>1</sup>

#### ABSTRACT

To understand the efficient machining of aerospace materials and achieve high quality machining, the application of coolants is essential and is based on the several factors including the types of machining process, workpiece material, tool nose radius, coolant concentration and cost. With the right type of coolants used, the performance of machining applications and the attributes of responses such as Surface roughness can be remarkably enhanced and HAZ can be improved. An attempt is made here to study the machinability and optimize the process parameters.

This paper presents an outline to optimize the Machinability studies on SAE AMS4413B alloy using Ethylene glycol as a coolant. The input parameters selected such as Cutting speed, Feed, depth of cut, coolant concentration, tool nose radius on Surface Roughness (Ra) and Heat Affected Zone (HAZ). A robust DOE based technique stresses for study of response variation using Central Composite design (CCD). The CCD design is a class of Response Surface Methodology used to design the matrix. The experiments are conducted for the confidence levels of 95%. R<sup>2</sup> predicted for Ra and HAZ is 95.26% and 95.51% respectively. As per the experimental analysis it is evident the individual effect of Cutting Speed, Feed, Depth of cut, Tool nose radius and Coolant concentration exists for the Ra and HAZ values. Optimum values of Ra and HAZ are 0.21 µm and 20.3 IACS. Selection of the Coolant concentration For Ethylene Glycol shall be in the range of 8-10%.

*Keywords:* Aluminum Lihium alloy - SAE AMS4413B, Ethylene Glycol, Central Composite Design (CCD) - Response surface methodology (RSM), Surface Roughness (Ra), Heat affected Zone (HAZ)

### 1. INTRODUCTION

Lightweight materials play a prominent role in the aerospace industry, a sector that has been a pioneer in adopting innovative materials and production technologies. Approximately four decades ago, aluminum emerged as a dominant material in aerospace, constituting up to 70% of an aircraft [1]. <sup>1</sup>Over time, the aerospace and space industry has been a catalyst for the development of new material systems and manufacturing methods. The primary drivers for these advancements include the imperative to reduce weight, enhance application-specific performance, and minimize costs [2]. The historical shift from wood to aluminum alloys as the main frame material dates back to the early 1920s [3]. The appeal of aluminum lies in its cost-effectiveness, lightweight nature, ability to attain high strength levels, and ease of fabrication. This unique combination makes aluminum alloys an attractive choice, with their cost-efficiency

<sup>&</sup>lt;sup>1</sup>Department of Mechanical Engineering, K. L. S. Gogte Institute of Technology, 590008, and Visvesvaraya Technological University, Belagavi, 590018, India.

<sup>\*</sup>Corresponding author

often directly linked to their ease of fabrication [3]

The initial development of Al–Li alloys can be traced back to research initiatives conducted by both the United States and the Soviet Union during the Cold War era. A notable outcome of these programs was the creation of the AA2020 alloy, patented by Alcoa in 1958 [4]. Thomas Dorian [5] highlighted the features of third-generation alloys, aligning with contemporary trends in transportation industries, especially aerospace, which emphasize enhancing performance, fuel efficiency, system lifespan, and environmental sustainability to create added value. The Pioneer Aviation companies like Airbus, Lockheed, Martin, Mikayan, Gurevich and Tupolev and others are aiming for the better improved technology under the lightweight materials. Considering their strength and density (Fig 1) it becomes quite obvious why aluminum and titanium alloys are the classical lightweight aerospace alloys.



Fig 1: Graph weight savings and property improvement. [6]

Al Li alloys of third generation used for aviation applications as shown in the below Table.1 [7]. Alloys 2195, 2196, 2297, 2397, 2198, 2099, 2199, 2050, 2060, and C99N were researched and developed for space and aircraft applications, and they are referred to as 3rd-generation Al-Li Products [8]. The next generation Al–Li alloys offers better weight savings less expensive to manufacture, operate and repair. The improved material properties improved damage tolerance performance. Al–Li alloys offer improvements in the fracture processes [9-10].

Al Li alloys	Li	Cu	Zn	Mg	Mn	Fe	Si	Cr	Zr	Ti	Other s
2050	0.7- 1.3	3.2- 3.9	0.25	0.2- 0.6	0.2- 0.5	0.1	0.0 8	0.0 5	0.06- 0.14	0.1	0.2- 0.7 Ag
2090	1.9- 2.6	2.4- 30.0	0.1	0.25	0.05	0.12	0.1	0.0 5	0.08- 0.15	0.1 5	
2098	0.8- 1.3	3.2- 3.8	0.35	0.25 -0.8	0.35	0.15	0.1 2		0.04- 0.18	0.1	0.25- 0.6 Ag
2099	1.6-	2.4-	0.4-	0.1-	0.1-	0.07	0.0	0.1-	0.05-	0.1	0.000

Table.1: Chemical Composition of Al-Li alloys [11]

	2.0	3.0	1.0	0.5	0.5		5	0.5	0.12		1 Be
2100	1.4-	2.0-	2.0-	0.05	0.1-	0.07	0.0		0.05-	0.1	0.000
2199	1.8	2.9	0.9	-0.4	0.5	0.07	5		0.12	0.1	1 Be

Machining is an essential part of any manufacturing industry to achieve desired components with accurate shape, size and surface finish through adequate cooling and lubrication [12]. Aluminum being an essential material being used in aerospace requirements [13]. The machining of Light weight alloys has been significant in recent years [14-27]. Study of coolants during machining to improve the surface finish and HAZ will be required for establishing the results and analysis of effect of machining parameters. Therefore, there is a need to implement sustainable cooling/lubrication system improves the machinability of light weight alloys [26]. The application of coolants is based on the several factors including the types of machining process, work-piece material, cutting tool and cost [27]. Many authors have represented their work using the various cooling lubrication method. In brief the different available cooling lubrication strategies.



Fig 2: Coolant strategies and methods for light weight alloys [28].

#### 2. LITERATURE REVIEW

The literature review deals with 2<sup>nd</sup> generation aluminum alloys as shown in Table 2 [29-41]. Various authors have presented the study of machinability characteristics on Al 7075 alloys. The responses evaluated suggest the use of cooling strategies, over Surface roughness, cutting force, temperature, chip morphology, etc. The authors [42-48] gave the scope on 3<sup>rd</sup> generation aluminum alloys shown in Table 3. Various authors discussed the machinability studies using the cooling strategies, for the responses such as the Surface Roughness, chip morphology, etc. Table 4. Represents the authors [49-53] having stated the use of ethylene glycol as a coolant for machinability studies.

A lot of experimental work have been done on  $2^{nd}$  generation materials. From the literature review, in the field of  $3^{rd}$  generation materials, responses of Surface roughness and Heat affected Zone using ethylene glycol as a coolant is still in experimental stage. In this investigation, DOE based RSM technique is employed to assess the output parameters. The conventional wet cooling has been used for machining of 3rd generation alloys to study the machinability characteristics of SAE AMS4413B.

Ref	Workpie ce	Cutting Environm	Machi ning	Parameter s evaluated Responses	Remarks
31	AA7075	Dry, MQL Cryogenic	Turning	CF, Cutting Temperatur e, chip Morpholog y, Tool Wear	<ul> <li>Feed component affected by cooling techniques.</li> <li>MQL and cryogenic led to lower temperature</li> <li>Cooling techniques affected the lesser Tool wear</li> </ul>
32	AA 7075-T6	Dry, MQL HPAJ, Cryogenic (LN2)(air, and vegetable oil emulsion)	Turning	Microstruct ure, Surface Roughness	<ul> <li>Analytical model for fatigue life prediction</li> <li>cryogenic technique led to lower temperature</li> </ul>
33	A17075	Dry and Cutting fluid using Gate ECM 1	Drilling	Tool wear, Burr Measureme nt	<ul> <li>Cutting fluid resulted in lower tool wear</li> <li>Cutting fluid resulted in 20% lower burr rate</li> </ul>
34	Al 7075	Dry	Turning	MRR, SR, Cylindricity error, circularity error	<ul> <li>Optimization of multiple outputs</li> <li>Increase in speed increased the CF also increase in SR and Circularity and cylindricity</li> </ul>
35	Al 7075	Dry	Milling	Temperatur e,	Behavior of different tool geometries and coatings
36	AA 7075	Dry, wet and cryogenic	Boring	SR, Temperatur e, Force	<ul> <li>Cryogenic leads to less Cutting force, temperature, Surface roughness</li> <li>Direct application of coolant recommended</li> </ul>
37	Al 7075 – T6	Dry, MQL, Cryogenic and Compresse d air	Turning	Tool wear, SR, Micro hardness, Force, Chip Morpholog y	<ul> <li>MQL, RHVT, and Compressed air, for improvement in SR and Tool wear</li> <li>Absence of Crater wear due to coolant conditions</li> </ul>
38	Al 7075 – T6	Coolant not specified	Milling	SR,	• Nose radius and depth of cut significant for minimizing the Surface roughness
39	Al 7075	Wet and LN2	Turning	SR, Temperatur e, Force	<ul> <li>Use of LN2 reduced the temperature, forces and surface roughness</li> </ul>

**Table 2:** 7075 Aluminum Alloy 2<sup>nd</sup> generation material

40	Al 7075	MQL, Compresse d Air	Milling	SR, MRR	<ul> <li>Speed and Feed rate was the important parameter that affected the SR</li> <li>Less viscous the coolant is effective</li> </ul>
41	Al 7075 – T6	Dry and MQL using Mecgreen 550 Lubricant	Turning	SR and Chip formation	<ul> <li>Chip formation is dependent on feed rate, CS and Lubricating conditions</li> <li>SR mainly dependent on feed rate</li> </ul>
42	Al 7136- T6511	Blastocut BC 35 SW	Milling	SR	• CS is the factor that Affects the SR.
43	Al 7075- T6	Nano Silver and Borax additives with 95% ethylene glycol (Suspensio n form)	Milling	SR, CF and Chip formation	<ul> <li>Tribological Performance of Borax and nano Silver added EG</li> <li>Nano Silver particles will improve the SR and solution prepared with EG has an industrial usage.</li> </ul>

 Table 3: 3<sup>rd</sup> Generation Aluminum alloys

Re	Workpi	Cutting	Mac	Paramete	Remarks
1.	/Tool material	ment	ng	rs evaluated Response s	
44	AA2024 -T3	Dry, Base Fluid (Mineral Oil) MQL, NFMQL MoS2	Turn ing	Surface Topograp hy, Temperat ure	<ul> <li>Surface Roughness was better with NFMQL than dry and MQL because of MoS2</li> <li>BUE is less with NFMQL</li> </ul>
45	AA2050	Soluble oil emulsion	Tapp ing	Machinin g Distortion MRR	<ul> <li>Optimization of Machining time</li> <li>Low values of distortion from the geometrical tolerances</li> </ul>
46	A12050	Dry and WET Emulsion Quakerco ol 7000 ALF 8%	Milli ng	Chip Morpholo gy	<ul> <li>CS assists in reducing the chip size</li> <li>Reduction in rake angle reduces the chip thickness</li> </ul>

47	2A 97	Dry	Milli ng	Chip Morpholo gy, Ra	<ul> <li>Cutting Speed played an important role for chip edge</li> <li>Convenient for replacing the Conventional Al alloys for aircrafts</li> </ul>
48	Al 8112	Vegetable oil (Copra Oil) as base oil, nano fluid	Milli ng	MRR	<ul> <li>Increase in depth of cut increases the MRR</li> <li>Helix angle plays an important role for improving the MRR</li> </ul>
49	Alumini um alloy 24345	Dry	Milli ng	Ra and Rz	<ul> <li>Uncoated and polished flute cutters perform far better coated cutters</li> <li>Rz and Sdr give a better picture of the actual surface from the functionality point of view</li> </ul>
50	Al 2024- T3	Dry	Drilli ng	Hole Deviation and Circularit y	<ul> <li>Burrs were less with uncoated drill bit</li> <li>CS also the significant parameter for geometrical parameters</li> </ul>

# Table 4: Coolant as Ethylene Glycol

Ref	Workpie	Cutting	Machin	Parameters	Remarks
•	ce	Environm	ing	evaluated	
		ent		Responses	
51	SUS 304	Cellulose nanocrystal ethylene glycol	Turning	Thermal Conductivity Analysis, cutting tool and chp thermal analysis,	• Thermal conductivity increases as concentration increases
52	AISI 1018	Ethylene glycol as base fluid, TiO2 and Al2O3 nanoparticl es	Turning	Temperature, SR	• Thermal Analysis was studied along with SR
	Al 3303	Water ethylene glycol mixture		Corrosion study	Corosion properties

53	AISI 304	Ethylene glycol TiO2 nano particles	Milling	Tool life, tool wear	<ul> <li>Tool ife decreases with increase increase in cutting speed</li> <li>Showed better performance with water soluble coolants in terms of tool life.</li> </ul>
54	AA6061- T6	Ethylene glycol mixture with TiO2 and ZnO hybrid nano coolant	Milling	SR, MRR	• Suitable alternative and compatible in using the hybrid nano-coolant
55	Fe3O4 nanofluid	Ehtylene glycol and waterbased mixture		Thermal Conductivity, Effect of temperture	Thermal Conductivity enhancement

### **3. EXPERIMENTAL SETUP**

All the machining has been carried out on an DMG MORI 3 Axis CNC milling machine. Machining was undertaken using three varying cutting parameters: cutting speed, feed rate and depth of cut. and coolant concentration with 8%, 10% and 12%. Suitable combinations of the cutting parameters were used to deduce the design of the experiment using the CCD method with three factors five levels Table 5. A total of 156 combinations of experiments were undertaken. The process parameters are as in the below table [57] The Milling was carried out with 100% engagement of 30 x 30 mm on the specimen. Aluminum Lithium alloy SAE AMS4413B slabs are used for experimentation. The size of each block was 50 x 50 x 20 mm as shown in the Fig 3. The chemical composition of the work piece material and the properties of the work piece material are shown in Table 6 and Table 7. The milling inserts (Fig. 4) by SECO Uncoated tools signature [59]. The uncoated insert is XOEX10T304 standard water-soluble Ethylene glycol coolant was used as the flood coolant as shown in Table 8.

Variables			Levels	
		-1	0	1
Cutting Speed(rpm)	X1	5000	11000	17000
Feed(mm/tooth)	X2	0.04	0.16	0.28
Depth of Cut(mm)	X3	0.5	1	1.5
Tool Nose Radius	X4	0.4	0.8	1.2
Coolant	X5	8	10	12
Concentration (%)				

 Table 5: Process parameters



Fig 3: Sample specimen drawing of size 50 x 50 x 20 mm

**Table 6:** Chemical composition of workpiece material.

Cu	Li	Mg	Mn	Ag	Zr	Bal.
3.5	0.9	0.40	0.35	0.45	0.12	Re m.
SAE A Solution Artific	M. SAE AMS4413B Solution Heat Treated, Stress Relieved, and Artificially Aged					

**Table 7:** Physical properties of 3rd Generation

Properties	Value
Density	270 gm/cc
Melting point	510 ° C
Tensile Strength	440 MPa
Yield Strength	370 MPa
Shear Strength	260 MPa
Fatigue Strength	125
Elastic Modulus	70-80 GPa
Poison's Ratio	0.33
Elongation	10%



#### Fig 4: Seco Insert Carbide Uncoated from XOEX10T3

Ethylene Glycol	1,2 Ethanediol
Ethylene Glycol	98%
Molecular	C2H6O2
formula	
Color	A clear colorless, slightly
	viscous liquid
Wt. per ML at	1.112-1.115g
20°C	

Table. 8: Coolant Details: Specifications of Ethylene Glycol: MOLYCHEM

#### **3.1 PERFORMANCE EVALUATION:**

Surface roughness and surface finish are opposite to each other, these are quantitative parameters. Surface roughness can be expressed in units of length after its measurement. "Measurement of finely spaced deviations of actual surface from nominal surface (datum) in the units of length ( $\mu$ m) is the measurement of surface roughness. Lesser the value of surface roughness better the surface finish is said. There are two popular methods of expressing measured value of surface roughness.

According to "AA" method surface roughness is the average of vertical deviations from the nominal surface over a specified surface length.

Average roughness (AA)

$$= \int_0^{lm} \frac{(Y)}{Lm} dx$$

Heat Affected Zone: Eddy current meter has been used to measure the HAZ. Portable meter with specifications High resolution color display and Accurate conductivity range: 0.5% IACS to 110% IACS, 0.28-64Ms/m, Digital Temperature readout The following pictures represent the experimental setup and testing done.



Fig: 5: Specimen Before machining



Fig. 6: Secimen After Machining





Fig. 7: Experimental set-up of the milling of a sample specimen



Surface roughness measurement using 410 Surf test on Sample specimen



HAZ measurment using Eddy current meter on Sample specimen

Fig 8: Testing of Ra and HAZ using 410 Surf-test and EDDY current meter

## 4. **RESULTS AND DISCUSSION**

In order to measure the statistical significance, analysis of variance (ANOVA) is used and

results for the Ra, and HAZ are tabulated in Table 9 and Table 10 respectively. The Results of interaction plots are placed.

The Regression equation of the analysis for Ra and Surface Roughness obtained is mentioned below:

**Ra** = 5.605 + 5.32 X2 - 1.674 X4 - 9.02 X2<sup>2</sup> + 0.0387 X5<sup>2</sup> + 0.000077 X1 \* X5 - 2.573 X2 \*X4 + 0.3649 X2 \*X5 - 0.0865 X3 \*X4 + 0.1102 X4 \*X5

 $\begin{array}{l} \textbf{HAZ} = 24.511 - 0.2131 \ X3 - 1.599 \ X4 - 0.582 \ X5 - 0.0909 \ X3^2 + 0.583 \ X4^2 + 0.02311 \ X5^{-2} \\ - 0.000063 \ X1 * \ X3 & + 0.000117 \ X1 * X4 - 0.000049 \ X1 & * X5 + 1.258 \ X2 & * X4 + 0.1430 \ X2 \\ * X5 - 0.0788 \ X3 & * X4 + 0.03185 \ X3 & * X5 + 0.07090 \ X4 & * X5 \end{array}$ 

							Signi fican
	Un-			Adj			t
	Coded		Adj	MS F-		Р	Fact
Source Ra	Values	DF	SS	Value	<b>F-Value</b>	value	ors
			37.695	1.8848		0.0000	
Model		20	80	0	13.70000	0	
			33.279	6.6559	479.2000	0.0000	
Linear		5	30	0	0	0	
			0.0644	0.0644		0.1380	
Cutting Speed	X1	1	0	0	4.63000	0	
			31.119	31.119	2240.510	0.0000	
Feed	X2	1	60	60	00	0	*
			0.0088	0.0088		0.3580	
Depth of Cut	X3	1	0	0	0.63000	0	
			1.2650	1.2650		0.0000	
Tool Nose Radius	X4	1	0	0	91.07000	0	*
Coolant			0.8217	0.8217		0.6760	
Concentration	X5	1	0	0	59.16000	0	
			1.0075	0.2015		0.0000	
Square		5	0	0	14.51000	0	*
Cutting Speed *			0.0245	0.0245		0.1860	
Cutting Speed	$X1^2$	1	0	0	1.76000	0	
			0.1253	0.1253		0.0030	
Feed * Feed	$X2^2$	1	0	0	0.02000	0	*
			0.0000	0.0000		0.9990	
DOC * DOC	X3 <sup>2</sup>	1	0	0	0.00000	0	
Tool Nose							
radius*Tool Nose			0.0402	0.0402		0.0910	
radius	$X4^2$	1	0	0	2.90000	0	
Coolant							
Concentration*Co							
olant			0.1750	0.1775		0.0000	
Concentration	X5 <sup>3</sup>	1	0	0	12.78000	0	*
2-Way			3.4090	0.3409		0.0000	
Interaction		10	0	0	24.54000	0	
Cutting	X1 *		0.0136	0.0136		0.3240	
Speed*Feed	X2	1	0	0	0.98000	0	

Cutting							
Speed*Depth of	X1 *		0.0052	0.0052		0.5410	
Cut	X3	1	0	0	0.38000	0	
Cutting							
Speed*Tool Nose	X1 *		0.0370	0.0370		0.1050	
radius	X4	1	0	0	2.66000	0	
Cutting							
Speed*Coolant	X1 *		0.2031	0.0203		0.0000	
Concentration	X5	1	0	1	14.63000	0	*
Feed*Depth of	X2 *		0.0878	0.0878		0.0130	
Cut	X3	1	0	0	6.32000	0	
Feed*Tool Nose	X2 *		1.4642	1.4642	105.4200	0.0000	
radius	X4	1	0	0	0	0	*
Feed*Coolant	X2 *		0.7364	0.7364		0.0000	
Concentration	X5	1	0	0	53.02000	0	*
Depth of							
Cut*Tool Nose	X3 *		0.1150	0.1150		0.0000	
radius	X4	1	0	0	8.28000	0	*
Depth of							
Cut*Coolant	X3 *		0.0000	0.0000		0.9600	
Concentration	X5	1	0	0	0.00000	0	
Tool Nose							
radius*Coolant	X4 *		0.7466	0.7466		0.0000	
Concentration	X5	1	0	0	53.75000	0	*

Table 9: ANOVA (Ra)

Table 10: ANOVA (HAZ)

	Un						Roopa K. F	ao et al. 905
	-						1	
	Со						Signifi	
a	ded	-		Adj	-	P	cant	
Source	Val	D	Adj	MS F-	F-	valu	Factor	
HAZ	ues	F	SS	Value	Value	e	S	
		2	9.62	0.400.60	143.51	0.00		
Model		0	100	0.48060	000	000		
<b>.</b> .		_	/.8/	1 57502	4/0.55	0.00		
Linear		3	910	1.5/583	000	000		
Cutting	X/1	1	0.04	0 105 47	31.500	0.00		
Speed	ΧI	1	055	0.10547	00	000		
Tee 1	vo	1	0.02	0.00104	6.2800	0.01	*	
Feed	ΛL	1	100	0.02104	0	300		
Danth of Cut	V2	1	6.93	6.02696	20/1.4	0.00		
Depin of Cut	Δ3	1	0.62	0.93080	196.66	000		
Tool Nose	V4	1	0.02	0 62510	180.00	0.00	*	
Caplant	Λ4	1	510	0.62510	000	000	-1-	
Concentratio			0.10		56 020	0.00		
Concentratio	V5	1	0.19	0 10067	50.950 00	0.00	*	
11	ЛЭ	1	070	0.19007	20 120	0.00		
Sauana		5	0.55	0.06720	20.120	0.00	*	
Cutting		5	000	0.00739	00	000		
Cutting								
Cutting			0.00		0.6800	0.40		
Speed	<b>X</b> 1 <sup>2</sup>	1	230	0.00229	0.0800	900		
Speed	Δ	1	0.00	0.00227	0.0100	0.94		
Feed * Feed	$\mathbf{X}2^2$	1	0.00	0.00002	0.0100	0.94	*	
	112	1	0.06	0.00002	18 330	0.00		
DOC * DOC	X3 <sup>2</sup>	1	140	0.06137	00	000	*	
Tool Nose	110	1	110	0.00137	00	000		
radius*Tool			0.06		19 290	0.00		
Nose radius	$X4^2$	1	460	0.06460	00	000	*	
Coolant		-		0.00100		000		
Concentratio								
n*Coolant								
Concentratio			0.06		18.950	0.00		
n	X5 <sup>3</sup>	1	350	0.06345	00	000	*	
2-Way		1	1.39		41.680	0.00		
Interaction		0	600	0.13960	00	000	*	
	X1							
Cutting	*		0.00		0.4100	0.52		
Speed*Feed	X2	1	140	0.00139	0	100		
Cutting	X1							
Speed*Dept	*		0.03		10.360	0.00		
h of Cut	X3	1	470	0.03469	00	200	*	
Cutting	X1							
Speed*Tool	*		0.01		5.6300	0.01		
Nose radius	X4	1	880	0.01884	0	900		
Cutting	X1		0.0-			0.07		
Speed*Cool	*	4	0.08	0.00100	24.300	0.00	ste	
ant	X5	1	140	0.08138	00	000	*	]

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Concentratio							
n							
	X2						
Feed*Depth	*		0.00		0.9200	0.33	
of Cut	X3	1	310	0.00309	0	800	
	X2						
Feed*Tool	*		0.34		104.46	0.00	
Nose radius	X4	1	980	0.34981	000	000	*
Feed*Coola							
nt	X2						
Concentratio	*		0.11		104.46	0.00	
n	X5	1	310	0.11309	000	000	*
Depth of	X3						
Cut*Tool	*		0.11		33.770	0.00	
Nose radius	X4	1	310	0.11309	00	000	*
Depth of							
Cut*Coolant	X3						
Concentratio	*		0.38		116.31	0.00	
n	X5	1	950	0.38951	000	000	*
Tool Nose							
radius*Cool							
ant	X4						
Concentratio	*		0.30		92.220	0.00	
n	X5	1	880	0.30883	00	000	*

## Interaction plots for Ra





radius e) Surface plot of Tool nose radius and Coolant connetration

Fig: 4.1.a-4.1.e shows the Surface plots when Ethylene glycol being used as the coolant with 8%, 10% and 12% concentration. The figures represent the effect of input parameters on the Surface Roughness (Ra). The Surface plots 4.1.a indicates the sudden decrease and increasing trend for cutting speed and coolant concentration. Whereas fig: 4.1.b and 4.1.c indicates the increasing trend for interaction between feed and coolant concentration as well as tool nose radius. The interaction plot 6.d indicates Tool nose radius and depth of cut a decreasing trend of Ra value. The Fig: 4.1.e shows a decreasing trend indicating higher value of tool nose radius and coolant concentration of 8-10%.

## Contour plots for Ra



and Tool Nose radius e) Contour plots for Coolant concentration and Tool nose radius

Fig 4.2.a – 4.2.e represents the contour plots for interaction between cutting speed, feed, depth of cut, tool nose radius and coolant concentration on Surface Roughness (Ra). The Fig 4.2.a shows the Simple maximum pattern of the contour plot for the hold values of speed 11000 rpm, 0.16 mm/tooth feed and Depth of cut of 1.5mm with Ra values less than 0.75 $\mu$ m. Fig: 4.2.b - 4.2.d represents the stationary ridge pattern and 4.2.e the rising ridge pattern on the contour plot with Ra values less than 0.50  $\mu$ m, 0.90  $\mu$ m respectively withhold values of speed 11000

rpm, 0.16 mm/tooth feed and Depth of cut of 1.5mm. From the above surface and contour plots it is observed that cutting speed between 11000 rpm and 17000 rpm can be considered with lower feed of 0.04 mm/tooth and larger tool nose radius and coolant concentration.



#### **Interaction plot for HAZ**



Fig: 4.3.a-4.3.e represents the Surface plots for Heat affected zone. The hold values for all the plots is with Cutting Speed as 11000 rpm, Depth of cut 1.5 mm, Tool nose radius as 0.8mm and 0.16 mm/tooth feed and 10% coolant concentration. Fig.6.a indicates a better lubrication since for all the speed values the HAZ is showing a decreasing trend for increasing Depth of cut. Fig. 4.b, indicates that the lower HAZ can be obtained by employing 0.4 mm tool nose radius with 10% coolant concentration. Fig 4.c shows sudden decrease and then increase in the HAZ for interaction between cutting speed and coolant concentration. Fig 3.d and 3.e indicates lower values of HAZ for cutting speed of 17000 rpm, lower feed and smaller tool nose radius along with 8%-10% coolant concentration. Fig.4.f and 4.g indicates that the Lower values of HAZ can be obtained with coolant concentration of 10% for dept of cut up to 2.5 mm. Fig. 4.6.h Shows lower HAZ for Tool nose radius as 0.8mm and 10% coolant concentration. From the surface plots up to 10% of coolant concentration and tool nose radius of 0.4-0.8 mm can be employed.

#### **Contour Plots for HAZ**





Tool nose radius, e) Contour plots for Feed and Coolant concentration, d) Contour plots for Depth of Cut and Tool nose radius, g) Contour plots for Depth of cut and Coolant concentration, h) Contour plots for Coolant concentration and Tool nose radius

Fig 4.4.a – 4.4.h represents the contour plots for interaction between cutting speed, feed, depth

of cut, tool nose radius and coolant concentration on Heat affected zone (HAZ). Fig.4.4.a indicates the stationary ridge pattern with HAZ values less than 20.3 IACS. Fig.4.4.f-4.4.g shows the minimax pattern of the contour plot for the hold values of speed 11000 rpm, 0.16 mm/tooth feed and Depth of cut of 1.5mm with HAZ values of 20. IACS. Fig: 3.4. b, c, d represents the stationary ridge pattern and 4.4.e the rising ridge pattern on the contour plot with HAZ values lesser than 20.5 and 20.6 IACS respectively withhold values of speed 11000 rpm, 0.16 mm/tooth feed and Depth of cut of 1.5mm. Fig 3.4.e indicates that the rising ridge pattern with HAZ 20.55 IACS value

### 5. CONCLUSION

Based on the experimental work the following conclusions were drawn:

The individual effect of Cutting Speed, Feed, Depth of cut, Tool nose radius and Coolant concentration exists for the Ra and HAZ values. Optimum values of Ra and HAZ are 0.21  $\mu$ m and 20.3 IACS. Selection of the Coolant concentration For Ethylene Glycol shall be in the range of 8-10%. The Linear model Analysis are well within the 95 % confidence level with R2 as 95.27 % for Ra values and 95.51 % for HAZ respectively. The experimental results shall be a scope for researchers.

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Fig 1: Graph weight savings and property improvement. [6]



Fig 2: Coolant strategies and methods for light weight alloys [28].



Fig.3: Sample specimen drawing of size 50 x 50 x 20 mm



Fig.4: Seco Insert Carbide Uncoated from XOEX10T3



Fig: 5: Specimen Before machining



Fig. 6: Secimen After Machining



Fig. 7: Experimental set-up of the milling of a sample specimen



Surface roughness measurement using 410 Surf test on Sample specimen



HAZ measurment using Eddy current meter on Sample specimen

Fig 8: Testing of Ra and HAZ using 410 Surf-test and EDDY current meter







**Migration Letters** 









**Fig:** 4.3.a) Surface plot of Cutting speed and depth of cut, b) Surface plot of Cutting speed and tool nose radius, c) Surface plot of CS and Coolant concentration, d) Surface plot of Feed and Tool nose radius, e) : Interaction plot: Surface plot of CS and Coolant connetration, f) Surface plot of Feed and Tool nose radius, g) Surface plot of Depth of cut and Coolant connetration, h) Surface plot of Coolant concentration and Tool nose radius





**Fig:** 4.3.a) Surface plot of Cutting speed and depth of cut, b) Surface plot of Cutting speed and tool nose radius, c) Surface plot of CS and Coolant concentration, d) Surface plot of Feed and Tool nose radius, e) : Interaction plot: Surface plot of CS and Coolant connetration, f) Surface plot of Feed and Tool nose radius, g) Surface plot of Depth of cut and Coolant connetration, h) Surface plot of Coolant concentration and Tool nose radius

Al Li alloys	Li	Cu	Zn	Mg	Mn	Fe	Si	Cr	Zr	Ti	Othe rs
2050	0.7- 1.3	3.2- 3.9	0.25	0.2- 0.6	0.2- 0.5	0.1	0.0 8	0.0 5	0.06- 0.14	0.1	0.2- 0.7 Ag
2090	1.9- 2.6	2.4- 30.0	0.1	0.25	0.05	0.12	0.1	0.0 5	0.08- 0.15	0.1 5	
2098	0.8- 1.3	3.2- 3.8	0.35	0.25 -0.8	0.35	0.15	0.1 2		0.04- 0.18	0.1	0.25- 0.6 Ag
2099	1.6- 2.0	2.4- 3.0	0.4- 1.0	0.1- 0.5	0.1- 0.5	0.07	0.0 5	0.1 - 0.5	0.05- 0.12	0.1	0.000 1 Be
2199	1.4- 1.8	2.0- 2.9	2.0- 0.9	0.05 -0.4	0.1- 0.5	0.07	0.0 5		0.05- 0.12	0.1	0.000 1 Be

Table.1: Chemical Composition of Al-Li alloys [11]

 Table 2: 7075 Aluminum Alloy 2<sup>nd</sup> generation material

Ref.	Workpiec	Cutting	Machini	Parameters	Remarks
	e	Environme	ng	evaluated	
		nt		Responses	
31	AA7075	Dry, MQL Cryogenic	Turning	CF, Cutting Temperature, chip Morphology, Tool Wear	<ul> <li>Feed component affected by cooling techniques.</li> <li>MQL and cryogenic led to lower temperature</li> </ul>

**Migration Letters** 

					• Cooling techniques affected the lesser Tool wear
32	AA 7075- T6	Dry, MQL HPAJ, Cryogenic (LN2)(air, and vegetable oil emulsion)	Turning	Microstructu re, Surface Roughness	<ul> <li>Analytical model for fatigue life prediction</li> <li>cryogenic technique led to lower temperature</li> </ul>
33	A17075	Dry and Cutting fluid using Gate ECM 1	Drilling	Tool wear, Burr Measurement	<ul> <li>Cutting fluid resulted in lower tool wear</li> <li>Cutting fluid resulted in 20%lower burr rate</li> </ul>
34	Al 7075	Dry	Turning	MRR, SR, Cylindricity error, circularity error	<ul> <li>Optimization of multiple outputs</li> <li>Increase in speed increased the CF also increase in SR and Circularity and cylindricity</li> </ul>
35	Al 7075	Dry	Milling	Temperature,	• Behavior of different tool geometries and coatings
36	AA 7075	Dry, wet and cryogenic	Boring	SR, Temperature, Force	<ul> <li>Cryogenic leads to less Cutting force, temperature, Surface roughness</li> <li>Direct application of coolant recommended</li> </ul>
37	Al 7075 – T6	Dry, MQL, Cryogenic and Compressed air	Turning	Tool wear, SR, Micro hardness, Force, Chip Morphology	<ul> <li>MQL, RHVT, and Compressed air, for improvement in SR and Tool wear</li> <li>Absence of Crater wear due to coolant conditions</li> </ul>
38	Al 7075 – T6	Coolant not specified	Milling	SR,	• Nose radius and depth of cut significant for minimizing the Surface roughness
39	A1 7075	Wet and LN2	Turning	SR, Temperature, Force	<ul> <li>Use of LN2 reduced the temperature, forces and surface roughness</li> </ul>
40	Al 7075	MQL, Compressed	Milling	SR, MRR	• Speed and Feed rate was the important parameter that affected

					• Less viscous the coolant is effective
41	Al 7075 – T6	Dry and MQL using Mecgreen 550 Lubricant	Turning	SR and Chip formation	<ul> <li>Chip formation is dependent on feed rate, CS and Lubricating conditions</li> <li>SR mainly dependent on feed rate</li> </ul>
42	Al 7136- T6511	Blastocut BC 35 SW	Milling	SR	• CS is the factor that Affects the SR.
43	Al 7075- T6	Nano Silver and Borax additives with 95% ethylene glycol (Suspension form)	Milling	SR, CF and Chip formation	<ul> <li>Tribological Performance of Borax and nano Silver added EG</li> <li>Nano Silver particles will improve the SR and solution prepared with EG has an industrial usage.</li> </ul>

 Table 3: 3<sup>rd</sup> Generation Aluminum alloys

Re f.	Workpie ce /Tool material	Cutting Environm ent	Mac hinin g	Paramete rs evaluated Responses	Remarks
44	AA2024- T3	Dry, Base Fluid (Mineral Oil) MQL, NFMQL MoS2	Turni ng	Surface Topograph y, Temperatu re	<ul> <li>Surface Roughness was better with NFMQL than dry and MQL because of MoS2</li> <li>BUE is less with NFMQL</li> </ul>
45	AA2050	Soluble oil emulsion	Tapp ing	Machining Distortion MRR	<ul> <li>Optimization of Machining time</li> <li>Low values of distortion from the geometrical tolerances</li> </ul>
46	A12050	Dry and WET Emulsion Quakerco ol 7000 ALF 8%	Milli ng	Chip Morpholo gy	<ul> <li>CS assists in reducing the chip size</li> <li>Reduction in rake angle reduces the chip thickness</li> </ul>
47	2A 97	Dry	Milli ng	Chip Morpholo gy, Ra	<ul> <li>Cutting Speed played an important role for chip edge</li> <li>Convenient for replacing the Conventional Al alloys for aircrafts</li> </ul>

48	Al 8112	Vegetable oil (Copra Oil) as base oil, nano fluid	Milli ng	MRR	<ul> <li>Increase in depth of cut increases the MRR</li> <li>Helix angle plays an important role for improving the MRR</li> </ul>
49	Aluminiu m alloy 24345	Dry	Milli ng	Ra and Rz	<ul> <li>Uncoated and polished flute cutters perform far better coated cutters</li> <li>Rz and Sdr give a better picture of the actual surface from the functionality point of view</li> </ul>
50	Al 2024- T3	Dry	Drilli ng	Hole Deviation and Circularity	<ul> <li>Burrs were less with uncoated drill bit</li> <li>CS also the significant parameter for geometrical parameters</li> </ul>

 Table 4: Coolant as Ethylene Glycol

Ref	Workpie	Cutting Environm	Machi ning	Parameters evaluated	Remarks
•		ent	ming	Responses	
51	SUS 304	Cellulose nanocrystal ethylene glycol	Turning	Thermal Conductivity Analysis, cutting tool and chp thermal analysis,	• Thermal conductivity increases as concentration increases
52	AISI 1018	Ethylene glycol as base fluid, TiO2 and Al2O3 nanoparticl es	Turning	Temperature, SR	Thermal Analysis was studied along with SR
	Al 3303	Water ethylene glycol mixture		Corrosion study	Corosion properties
53	AISI 304	Ethylene glycol TiO2 nano particles	Milling	Tool life, tool wear	<ul> <li>Tool ife decreases with increase increase in cutting speed</li> <li>Showed better performance with water soluble coolants in terms of tool life.</li> </ul>

54	AA6061- T6	Ethylene glycol mixture with TiO2 and ZnO hybrid nano coolant	Milling	SR, MRR	• Suitable alternative and compatible in using the hybrid nano-coolant
55	Fe3O4 nanofluid	Ehtylene glycol and waterbased mixture		Thermal Conductivity, Effect of temperture	Thermal Conductivity enhancement

## Table 5: Process parameters

Variables			Levels	
		-1	0	1
Cutting Speed(rpm)	X1	5000	11000	17000
Feed(mm/tooth)	X2	0.04	0.16	0.28
Depth of Cut(mm)	X3	0.5	1	1.5
Tool Nose Radius	X4	0.4	0.8	1.2
Coolant	X5	8	10	12
Concentration (%)				

**Table 6:** Physical properties of 3rd Generation

Cu	Li	Mg	Mn	Ag	Zr	Bal.		
3.5	0.9	0.40	0.35	0.45	0.12	Re m.		
SAE AMS4413B Solution Heat Treated, Stress Relieved, and Artificially Aged								

# Table:7: Physical properties of 3rd Generation

Properties	Value
Density	270 gm/cc
Melting point	510 ° C
Tensile Strength	440 MPa
Yield Strength	370 MPa
Shear Strength	260 MPa

Fatigue Strength	125
Elastic Modulus	70-80 GPa
Poison's Ratio	0.33
Elongation	10%

 Table. 8: Coolant Details: Specifications of Ethylene Glycol: MOLYCHEM

Ethylene Glycol	1,2 Ethanediol
Ethylene Glycol	98%
Molecular	C2H6O2
formula	
Color	A clear colorless, slightly
	viscous liquid
Wt. per ML at	1.112-1.115g
20°C	-

Table 4.1: ANOVA (Ra)

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Source Itu	( undeb	<i>D</i> 1	37.6958	( unde	1 (4140	1 vulue	
Model		20	0	1.88480	13.70000	0.00000	
			33.2793				
Linear		5	0	6.65590	479.20000	0.00000	
Cutting Speed	X1	1	0.06440	0.06440	4.63000	0.13800	
			31.1196	31.1196			
Feed	X2	1	0	0	2240.51000	0.00000	*
Depth of Cut	X3	1	0.00880	0.00880	0.63000	0.35800	
Tool Nose Radius	X4	1	1.26500	1.26500	91.07000	0.00000	*
Coolant							
Concentration	X5	1	0.82170	0.82170	59.16000	0.67600	
Square		5	1.00750	0.20150	14.51000	0.00000	*
Cutting Speed *	*** 2		0.00450	0.00450	1	0.40.600	
Cutting Speed	$Xl^2$	1	0.02450	0.02450	1.76000	0.18600	
Feed * Feed	$X2^2$	1	0.12530	0.12530	0.02000	0.00300	*
DOC * DOC	X3 <sup>2</sup>	1	0.00000	0.00000	0.00000	0.99900	
1001 Nose							
radius	$\mathbf{X}A^2$	1	0.04020	0.04020	2 90000	0.09100	
Coolant	744	1	0.04020	0.04020	2.90000	0.07100	
Concentration*Coola							
nt Concentration	X5 <sup>3</sup>	1	0.17500	0.17750	12.78000	0.00000	*
2-Way Interaction	110	10	3.40900	0.34090	24.54000	0.00000	
Cutting Speed*Feed	X1 * X2	1	0.01360	0.01360	0.98000	0.32400	
Cutting Speed*Depth							
of Cut	X1 * X3	1	0.00520	0.00520	0.38000	0.54100	
Cutting Speed*Tool							
Nose radius	X1 * X4	1	0.03700	0.03700	2.66000	0.10500	
Cutting							
Speed*Coolant	X1 * X5	1	0.20310	0.02031	14.63000	0.00000	*
					Μ	ligration Lette	ers

Concentration							
Feed*Depth of Cut	X2 * X3	1	0.08780	0.08780	6.32000	0.01300	
Feed*Tool Nose							
radius	X2 * X4	1	1.46420	1.46420	105.42000	0.00000	*
Feed*Coolant							
Concentration	X2 * X5	1	0.73640	0.73640	53.02000	0.00000	*
Depth of Cut*Tool							
Nose radius	X3 * X4	1	0.11500	0.11500	8.28000	0.00000	*
Depth of							
Cut*Coolant							
Concentration	X3 * X5	1	0.00000	0.00000	0.00000	0.96000	
Tool Nose							
radius*Coolant							
Concentration	X4 * X5	1	0.74660	0.74660	53.75000	0.00000	*

Table 4.2 : ANOVA (HAZ)

	Un-					Roopa	K. Rao et al. 93
	cod ed Valu	D	Adj	Adj MS		Р	Significa nt
Source HAZ	es	F	SS	<b>F-Value</b>	<b>F-Value</b>	value	Factors
			9.621		143.5100	0.0000	
Model		20	00	0.48060	0	0	
			7.879		470.5500	0.0000	
Linear		5	10	1.57583	0	0	
			0.040			0.0000	
Cutting Speed	X1	1	55	0.10547	31.50000	0	
			0.021			0.0130	
Feed	X2	1	00	0.02104	6.28000	0	*
			6.936		2071.400	0.0000	
Depth of Cut	X3	1	90	6.93686	00	0	
Tool Nose			0.625		186.6600	0.0000	
Radius	X4	1	10	0.62510	0	0	*
Coolant			0.190			0.0000	
Concentration	X5	1	70	0.19067	56.93000	0	*
			0.330			0.0000	
Square		5	00	0.06739	20.12000	0	*
Cutting Speed *			0.002			0.4090	
Cutting Speed	X1 <sup>2</sup>	1	30	0.00229	0.68000	0	
			0.000			0.9400	
Feed * Feed	$X2^2$	1	00	0.00002	0.01000	0	*
			0.061			0.0000	
DOC * DOC	X3 <sup>2</sup>	1	40	0.06137	18.33000	0	*
Tool Nose							
radius*Tool			0.064			0.0000	
Nose radius	X4 <sup>2</sup>	1	60	0.06460	19.29000	0	*
Coolant							
Concentration*							
Coolant			0.063			0.0000	
Concentration	X5 <sup>3</sup>	1	50	0.06345	18.95000	0	*
2-Way			1.396			0.0000	
Interaction		10	00	0.13960	41.68000	0	*
Cutting	X1 *		0.001			0.5210	
Speed*Feed	X2	1	40	0.00139	0.41000	0	
Cutting							
Speed*Depth of	X1 *		0.034			0.0020	
Cut	X3	1	70	0.03469	10.36000	0	*
Cutting							
Speed*Tool	X1 *		0.018			0.0190	
Nose radius	X4	1	80	0.01884	5.63000	0	
Cutting							
Speed*Coolant	X1 *		0.081			0.0000	
Concentration	X5	1	40	0.08138	24.30000	0	*
Feed*Depth of	X2 *		0.003			0.3380	
Cut	X3	1	10	0.00309	0.92000	0	
Feed*Tool Nose	X2 *		0.349		104.4600	0.0000	
radius	X4	1	80	0.34981	0	0	*
Feed*Coolant	X2 *		0.113		104.4600	0.0000	
Concentration	X5	1	10	0.11309	0	0	*

Depth of							
Cut*Tool Nose	X3 *		0.113			0.0000	
radius	X4	1	10	0.11309	33.77000	0	*
Depth of							
Cut*Coolant	X3 *		0.389		116.3100	0.0000	
Concentration	X5	1	50	0.38951	0	0	*
Tool Nose							
radius*Coolant	X4 *		0.308			0.0000	
Concentration	X5	1	80	0.30883	92.22000	0	*